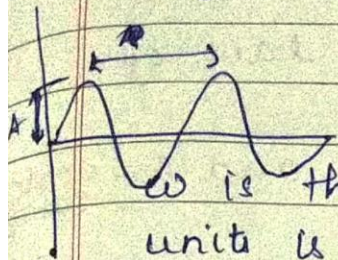


Amplitude: The maximum displacement A , is known as Amplitude. The period T , is the time for one revolution, since the amplitude angular path for one revolution is 2π rad.



$$T = \frac{2\pi \text{ rad}}{\omega} \quad - (1)$$

ω is the angular momentum velocity, its SI units is rad s^{-1} . The **frequency** is the reciprocal of the period

$$f = \frac{1}{T} = \frac{\omega}{2\pi \text{ rad}} \quad - (2)$$

$$\omega = (2\pi \text{ rad}) f$$

The mathematical form of displacement y is shown as

$$y = A \sin(\omega t) \quad - (3)$$

where ωt is known as phase.

On first differentiation

$$\frac{dy}{dt} = \omega A \cos \omega t \quad - (4)$$

On second diff. $\frac{d^2y}{dt^2} = -A\omega^2 \sin \omega t \quad - (5)$

hence $\boxed{\frac{d^2y}{dt^2} = -\omega^2 y} \quad - (6)$

This equation is known as simple harmonic motion equation.

Also according to Hooke's law,

$$F = -k_n y \quad \text{--- (7)}$$

where F is restoring force of a spring and y a displacement.

k_n is force constant.

According to Newton's second law of motion.

$$F = m \cdot a \quad (m = \text{mass and } a = \text{acceleration})$$

$$\therefore F = m \frac{d^2 y}{dt^2} \quad \text{--- (8)}$$

On equating eq. (6) and (8)

$$\frac{d^2 y}{dt^2} = \frac{F}{m} = -\omega^2 y$$

$$\text{or } -\frac{k_n y}{m} = -\omega^2 y \quad \text{--- (9)}$$

$$\text{or } \omega = \sqrt{\frac{k_n}{m}}$$

$$\Rightarrow \nu = \frac{1}{2\pi} \sqrt{\frac{k_n}{m}} \quad \text{--- (10)}$$

This frequency is referred as frequency of simple harmonic motion.

The Basics of Quantum mechanics

Schrodinger's theory often referred as wave mechanics was based on the realization of wave properties of elementary particles such as electrons.

Using the Compton effect it was deduced that

$$\lambda = \frac{h}{p} = \frac{h}{mv}$$

Since we know that electrons have wave properties

$$2\pi r = n\lambda$$

$$\text{or } 2\pi r = \frac{nh}{mv}$$

$$\text{or } mvr = \frac{nh}{2\pi}$$

Here mvr is the orbital angular momentum which according to Bohr is an integral number of $\frac{h}{2\pi}$.

The Uncertainty principle:-

It states that if we avoid the momentum change using radiation of long wavelength then we won't be able to determine the change in position precisely.

Mathematically,

uncertainty in position Δq \rightarrow uncertainty in momentum Δp

$$\Delta q \cdot \Delta p \approx \frac{h}{4\pi} \approx \frac{1}{2} \frac{h}{2\pi}$$

$\left\{ \frac{h}{4\pi} = \frac{h}{2\pi} \right\}$

Thus the product of these uncertainty decreases as the mass increases.

For kinetic factor

$$\Delta t = \frac{\Delta q}{u}$$

Since energy is $\frac{1}{2}mu^2$ and momentum is mu , the uncertainty in energy is given by

$$\Delta E = u \Delta p$$

Hence the eq. becomes

$$\Delta E \cdot \Delta t \approx \frac{h}{4\pi} \approx \frac{1}{2} \frac{h}{2\pi}$$

Schrodinger's wave mechanics

For a wave travelling in vacuum with speed c , the wave equation is

$$\frac{\partial^2 y}{\partial t^2} = \frac{1}{c^2} \frac{\partial^2 y}{\partial x^2} \quad \text{--- (1)}$$

where y is displacement in y -direction
Schrodinger replaced this equation with:

$$\frac{\partial^2 \psi}{\partial x^2} = \frac{1}{c^2} \frac{\partial^2 \psi}{\partial t^2} \quad \text{--- (2)}$$

$\psi(x, t)$ is the function that describes the behaviour of electrons and can be written as

$$\psi(x, t) = Ce^{i\alpha}$$

$$\text{where } i = \sqrt{-1}$$

$$\text{and } \alpha = 2\pi \left(\frac{x}{\lambda} - \nu t \right)$$

$\psi(x, t)$ is known as eigen function or wave function and can be split into

$$\psi(x, t) = \psi(x) e^{-2\pi i \nu t}, \quad \psi(x) = e^{2\pi i x/\lambda}, \quad e^{-2\pi i \nu t}$$

where $\psi(x)$ is a function of x , not t
and $e^{-2\pi i \nu t}$ is a function of t , not x .

On replacing v by E/h and λ by h/p we get

$$\psi(x,t) = C e^{2\pi i x p_x / h} e^{-2\pi i E t / h}$$

On partial differentiation with respect to t we get.

$$\frac{\partial \psi(x,t)}{\partial t} = -\frac{2\pi i E C}{h} e^{2\pi i x p_x / h} e^{-2\pi i E t / h}$$
$$= -\frac{2\pi i E}{h} \psi(x,t)$$

On rearranging the eqn.

$$\frac{-h}{2\pi i} \frac{\partial \psi}{\partial t} = E \psi$$

This is an operator equation and operator \hat{H} is written as

$$\frac{-h}{2\pi i} \frac{\partial}{\partial t}$$

For differentiating the same eqn with respect to x we get

$$\frac{\partial \psi(x,t)}{\partial x} = \frac{2\pi i p_x}{h} \psi(x,t)$$

Rearranged to

$$\frac{h}{2\pi i} \frac{\partial \psi}{\partial x} = p_x \psi \quad \Rightarrow \quad \hat{p}_x = \frac{h}{2\pi i} \frac{\partial}{\partial x}$$

\hat{p}_x is operator for momentum

For total energy of a system
 $E_{\text{total}} = E_{\text{kinetic}} + E_{\text{potential}}$
 $= \frac{p^2}{2m} + E_p(x,t)$

E_{pot}

on expressing E in term of momentum it is called **Hamiltonian** and the eqn becomes.

$$\hat{H} = \frac{-\hbar^2}{8\pi^2 m} \frac{\partial^2}{\partial x^2} + E_p(x,t)$$

and overall equation is

$$\left[\frac{-\hbar^2}{8\pi^2 m} \frac{\partial^2}{\partial x^2} + E_p(x,t) \right] \psi = \frac{-\hbar}{2\pi i} \frac{\partial \psi}{\partial t}$$

The operator $\frac{\partial^2}{\partial x^2}$ can be replaced by

$$\nabla^2 = \frac{\partial^2}{\partial x^2} + \frac{\partial^2}{\partial y^2} + \frac{\partial^2}{\partial z^2}$$

∇^2 is the del-squared operator or **Laplacian operator**.

hence the equation becomes

$$\left[\frac{-\hbar^2}{8\pi^2 m} \nabla^2 + E_p(x,y,z) \right] \psi = \frac{-\hbar}{2\pi i} \frac{\partial \psi}{\partial t}$$

Time-dependent Schrodinger wave equation.